

# Supercapacitors and electrochemical energy storage





## Overview

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Are supercapacitors a good energy storage device?

Abstract. Supercapacitors are among the most promising electrochemical energy-storage devices, bridging the gap between traditional capacitors and batteries in terms of power and energy density. Their charge-storage performance is largely influenced by the properties of electrode materials, electrolytes and the underlying charge-storage mechanisms.

Are rechargeable batteries and supercapacitors a good choice for electrochemical energy storage?

As a result, there has been a great interest in developing efficient electrochemical energy storage (EES) devices. Among EES technologies, rechargeable batteries (RBs) and supercapacitors (SCs) are the two most desired candidates for powering a range of electrical and electronic devices [3, 4, 5, 6, 7, 8, 9, 10].

How are supercapacitors classified based on the charge storage mechanism?

Supercapacitors are classified based on the charge storage mechanism into two primary types: electrochemical double-layer capacitors (EDLC) and pseudocapacitors . Understanding the charge storage mechanisms in these two types of supercapacitors is crucial for comprehending supercapacitors' unique characteristics and applications. 2.1.1.

Do supercapacitors have a high energy density?

1) The energy densities of electrochemical capacitors are not high. Currently, there remains a noticeable gap between the energy densities of supercapacitors ( $<20 \text{ Wh kg}^{-1}$ ) and batteries ( $30\text{--}200 \text{ Wh kg}^{-1}$ ). [474 - 476] Improving energy storage density continues to be a key research focus and challenge in the field of supercapacitors.



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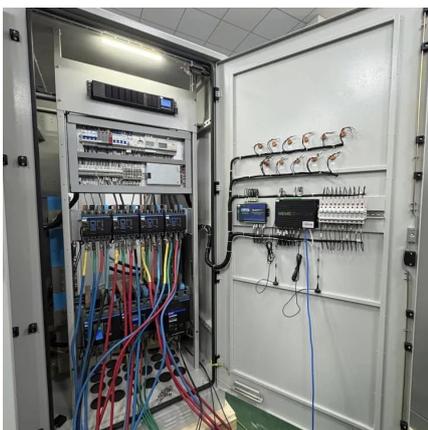
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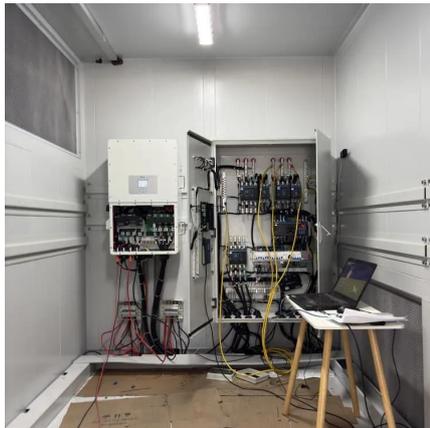
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